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**perform the**

**catalyzed**

**decomposition of**

**hydrogen peroxide**

**under various**

**conditions We will**

**record the trials**

**using a gas**

**pressure sensor in a**

**lab quest and**

**analyze graphs of**

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**the data From this analysis we will determine the rate constant the activation energy and the rate law'**

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burette with 50 ml of  
0.2M potassium  
permanganate  
solution add 10 ml'**

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and 2 mL of 6M  
sulfuric acid to a Lab  
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