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though geochemistry

is a subdiscipline

of earth science it

is a very broad

topic so broad in

fact that no one can

really master it all

geochemists

invariably

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specialize in one or  
a few aspects such  
as geochemical  
thermodynamics  
isotope geochemistry  
marine chemistry  
atmospheric  
chemistry trace  
element geochemistry  
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nuclear origins it  
then reviews  
radioactive decay  
schemes and their  
use in geochronology

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a following chapter covers the closely related techniques such as fission track and carbon 14 dating subsequent chapters cover nucleosynthetic anomalies'

'geochemistry'  
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geochemistry is the science that uses the tools and principles of chemistry to explain the mechanisms behind major geological systems such as the earth's crust and its oceans  
1 the realm of geochemistry extends

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beyond the earth  
encompassing the entire  
solar system and has  
made important  
contributions to the  
understanding of a  
number of processes  
including mantle  
convection the  
formation of  
isotopes  
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agu and has been  
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author by isi

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stable isotope  
geochemistry the  
following is a brief  
review of some of  
the fundamentals of  
stable isotope  
geochemistry  
including  
definitions  
terminology basic  
principles standards

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and guidelines on  
reporting data the  
sources of the text  
are given at the  
end'

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nucleus 2 1 3  
radioactive decay 12  
1 4 nucleosynthesis  
16 chapter 2 decay  
systems and  
geochronology i 32 2

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1 basics of  
radioactive isotope  
geochemistry 32 2 2  
geochronology 36 2 3  
the k ar ca system  
39 2 4 the rb sr  
system 47 2 5 the sm  
nd system 50 2

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- isotope  
fractionation  
factors play a key  
role in modern  
geochemistry and are  
used to interpret a  
broad range of  
natural phenomena  
over a wide range of  
temporal scales  
experimental  
advances in this

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area have been  
driven by  
significant  
improvements in mass  
spectrometry  
techniques coupled  
to advances in  
computational chemistry  
methods due to  
advances in software  
and substantial  
improvements in'

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applications

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text containing 83  
tables and 181  
figures covers a  
wide variety of  
topics ranging from  
atomic structure to

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chemical and  
isotopic equilibria  
to modern  
biogeochemical  
cycles which are  
divided into four  
interrelated parts  
crystal chemistry  
chemical reactions  
and biochemical  
reactions involving  
bacteria isotope

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geochemistry  
radiogenic and  
stable isotopes and  
the earth  
supersystem'  
' isotope  
fractionation

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isotope  
fractionation  
describes processes  
that affect the

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relative abundance  
of isotopes often  
used in isotope  
geochemistry  
normally the focus  
is on stable  
isotopes of the same  
element isotopic  
fractionation in the  
natural environment  
can be measured by  
isotope analysis

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using isotope ratio  
mass spectrometry to  
separate different  
element isotopes on  
the basis of their  
mass to charge  
ratio'

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there has been an explosion of stable isotope data in the literature over the past few years establishment of routine analytical techniques and advances in mass spectrometry have made entrance into this field far less

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formidable than in  
former years today  
there are about 150  
laboratories  
throughout the world  
where research in  
stable isotope  
geochemistry is  
conducted in august  
1976 a'

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isotopes in  
carbonates are a  
rapidly developing  
and promising  
paleothermometry  
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oxygen isotope ? 18  
o thermometry the  
temperature  
estimates they  
provide are  
independent of the  
isotopic position of  
the water source in  
which the carbonate  
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second most abundant  
element in the earth  
s crust and is an  
important nutrient  
in the ocean the  
global si cycle  
plays a critical  
role in regulating  
primary productivity  
and carbon cycling

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on the continents  
and in the oceans  
development of the  
analytical tools  
used to study the  
sources sinks and  
fluxes of the global  
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modified hydrous  
version can be found  
in the works of  
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11 The Mantle and  
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mantle and core 487  
 $d = \rho \frac{dr}{g r^2} = \rho \frac{v}{p}$   
 $\frac{2}{3} \frac{v}{s} = \frac{2}{3} \frac{4}{3} \rho r$   
 $r^2 dr = \frac{11}{12} \text{ equation}$   
11 12 describes how  
density changes in a  
self compressing but  
otherwise uniform  
sphere and is known  
as the Adams

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was recognized by  
geologists as a  
useful tool in  
petroleum

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exploration but  
further progress was  
impeded by several  
unresolved problems  
1 inadequate  
background on the  
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biological molecules

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precursors of  
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mantle 161 6 1  
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balancing depleted

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*mantle and crust 172*

*6 4 mantle plume*

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**'applications of  
stable isotopes in  
clinical**

**pharmacology**

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**isotope ratio mass  
spectrometry**

**originates from**

**geochemistry where**

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minor differences of isotopes at the natural abundance level are to be detected to determine the origin of for instance fossils an advantage of IRMS is the high precision for the isotope ratio measurement'

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geochemists provided  
the earliest  
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the tracing of water  
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added storeys to the  
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systematics and  
geochemistry of a  
plume***

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northeastern india  
underwent early  
cretaceous basic and*

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alkaline carbonatite  
activity many  
earlier workers  
proposed spatial and  
temporal association  
of these magmatisms  
with a plume  
currently beneath  
kerguelen storey et  
al 1992 kent et al  
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*al 2005 it is  
suggested that  
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earth the first part

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the tools of

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isotope geochemistry  
takes advantage of  
naturally occurring*

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variations in the isotopic position of elements resulting from natural nuclear and chemical processes to further understand earth processes both in the near surface environment and the earth's deep interior'

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geochemistry of the  
galapagos islands**  
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hf isotope ratios  
thus confirm the  
major features of  
galapagos isotope  
geochemistry  
observed by earlier  
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homogeneous isotopic

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positions on the  
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scale of the entire  
archipelago and 3  
the most depleted  
isotopic signatures  
in a central region  
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parative studies  
involving earth  
materials have so  
far focussed on  
characterizing  
overall variance in  
nd stable isotope

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concentrations and  
the  $^{145}\text{Nd}$  and  
 $^{146}\text{Nd}$  of mantle and  
igneous crustal 'gas  
**source mass**  
**spectrometry stable**  
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**geochemistry is used**  
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that applications  
span all fields of  
the geosciences  
variations in bond  
strength and related  
variability in  
molecular  
vibrational modes  
lead to mass  
dependent isotopic  
fractionations among  
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