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structures which show
how the valence
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in the molecule There
are also many examples
of these along with
notable exceptions so
we will not belabor the
point here Becoming
proficient at drawing
Lewis structures
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sour and cause certain

dyes to change color

?Bases taste bitter and

feel soapy'

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Bonding Francis Howell
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Bonding 1 A chemical
bond represents a force
that holds groups of two
or more atoms together
and makes them
function as a unit 2 An
ionic compound results
when a metallic element
reacts with a nonmetallic
element An example is
the reaction of the metal
sodium with the
nonmetal chlorine $2\text{Na} + \text{s}$
 $\text{Cl}_2 \text{g} \rightarrow 2\text{Na} \text{Cl} \text{s}$ 3'**

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properties depends on

Pressure P Is equal to

Pressure P SI unit"para 1

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These include alcohols primary secondary and tertiary aldehydes

ketones carboxylic acids and esters Naming these

compounds is quite

systematic and

presented quite clearly

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concentration of a
reactant or product per
unit of time t A t t
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