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Electrochemistry Elijah Harris Two cells were combined by using an alligator clip to attach the Copper in one cell to the Zinc'

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cells Name Narynbek

Gilman burns mouth and

wearing lab coats

Discussion 1 $Zn(s) + Cu^{2+}(aq)$

$Cu(s)$

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Laboratory Goals In this

laboratory you will $\frac{3}{4}$

develop a basic

understanding of what

electrochemical cells are $\frac{3}{4}$

develop familiarity with a

few different examples of

redox reactions'

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Shorthand cell designation

1 Zn Cu²⁺ || Zn²⁺ Cu²⁺

Sn || Sn²⁺ Cu'

'Experiment 12

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Report Experiment 12

from CHEM 1BL 1BL at

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Electrochemical Cells This

experiment familiarized us

with the Nernst Equation

in order to find ΔG , ΔH and

"Experiment

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Introduction In this lab

you will be able to observe

changes when an electric

current passes through a

solution In all

electrochemical cells

electrons move through the

wire the external'

'Experiment 9

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Galvanic Cell

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Galvanic Cell Introduction provided at the end of the In Lab section For the reduction half reactions in equations 2'

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